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Synthesis of AB Ring Model System of Taxol via Allylation of 8-Membered Ring Compound and Intramolecular Aldol Condensation

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1-t-Butyldimethylsiloxy-8,11,11-trimethylbicyclo[5.3.1]undec -7-en-9-one (2) has been synthesized via intramolecular aldol condensation of the precursor 3 by combined use of lithium disopropylamide (LDA) and CeCl₃. The dicarbonyl compound 3 was prepared by allylation of 8-membered ring ketone 6b.

A useful method for the synthesis of optically active 8-membered ring enone (1), a new and potential synthetic intermediate of taxol, ¹ by way of stereoselective aldol reactions and intramolecular Reformatsky-type reaction was reported in previous communication.² In our synthetic strategy, AB ring system of taxol is to be constructed by introducing A ring segment onto 8-membered ring compound 1, followed by intramolecular cyclization of the precursor as shown in Scheme 1.

House et al. reported an effective method for the synthesis of bicyclo[5.3.1]undec-7-en-9-one skeleton by way of intramolecular aldol-type condensation in the presence of sulfuric acid and acetic acid in 1980.³ Further, Swindell et al. described constructions of AB ring system of taxinine through aldol-type condensation carried out under basic conditions in 1990.⁴ In these cases, no oxygenated functions were involved at C-1 position of the synthesized cyclic compounds; therefore, a development of convenient method for the synthesis of 2 which possesses an alkoxy function at C-1 position from 8-membered ring compound 6 is strongly desired in order to complete total synthesis of taxol from 8-membered ring compound 1.

In the first place, starting material **6a** or **6b** was prepared from the known dicarbonyl compound 7^5 by way of methylation, mono-reduction and protection of resulting alcohol. Though aldol

reactions between **6a** or **6b** and the lithium enolate of methyl ethyl ketone, ethyl acetate or ethyl propionate, were attempted in order to introduce carbonyl unit onto 8-membered ring compound, these aldol-type reactions did not proceed at all. To this end, two routes including alkylations of **6** and the subsequent oxygenations were planned as a model synthesis of taxol (Scheme 1, Paths I and II).

Path I: alkylation of 6a using vinyl cerium reagent proceeded smoothly to give a mixture of siloxy alcohols in good yield (95%, cis/trans = 59/41).6 In attempted protections of resulting allyl alcohols, no desired product was formed because of steric hindrances of the alcohols. On the other hand, cyclic silyl ether 9 was unexpectedly produced when cis-siloxy alcohol 8 was treated with NaH. The corresponding keto t-butyldimethylsilyl ether 4 was obtained by regioselective alkylation of cyclic silyl ether 9 with t-butyllithium followed by Swern oxidation. Since hydroboration of 4 did not work at all due to bulkiness of the substrate, hydroboration of the cis-siloxy alcohol 8 was investigated at first, and the corresponding diol was formed with good regioselectivity. Silyl ether 10a was produced by regioselective protection of the diol, then cyclic silyl ether 11 was obtained by deprotection of silyl ether 10a and protection of resulting cis-diol. Unfortunately, alkylation of 11 with t-butyl lithium gave undesired silyl ether 10b exclusively.

a) K₂CO₃, MeI, acetone, r.t. (97%); NaBH₄, EtOH, r.t. (84%); (6a) PhMe₂CMe₂SiCl, Imidazole, DMAP, DMF, r.t. (99%); (6b) TBSCl, Imidazole, DMAP, DMF, r.t. (97%); b) CH₂=CHMgBr, CeCl₃, THF, -78 °C (95%, cis / trans = 59 /41); c) NaH, TESCl, DMF, 0 °C to r.t. (50%); d)^tBuLi, Et₂O, -78 °C (65%); (COCl)₂, DMSO, Et₃N, CH₂Cl₃, -78 °C to r.t. (98%); e) ThexBH₂, THF; 9N NaOHaq., 30% H₂O₂aq. (80%); NaH, BnBr, THF, 0 °C (97%); f) TBAF, THF, r.t. (99%); Me₂SiCl₂, Imidazole, DMF, r.t. (84%); g)^tBuLi, Et₂O, -78 °C (82%)

Scheme 2.

Path II: then, a route including allylation of 8-membered ring ketone, followed by Wacker oxidation was examined. Allylation of 6b with allylmagnesium bromide took place smoothly to produce a mixture of siloxy alcohols in excellent yield (97%, cis/trans = 53/47). The following treatments of cis-siloxy alcohol, e.g. deprotection of silyl ether and protection of thus formed cisdiol, gave cyclic silyl ether 12. Further, alkylation of 12 with tbutyllithium, followed by Swern oxidation gave desired 8-membered ring ketone 5. Successive Wacker oxidation of the terminal olefin of 5 and selective methylation of the lithium enolate derived from the diketone 13 afforded dicarbonyl compound 3 in good yield.

a) allyIMgBr, THF: $Et_2O=3:1,0$ °C (97%, cis/trans = 53/47); TBAF, THF, 0 °C (100%); Me_2SiCl_2 , Imidazole, DMF, r.t. (89%); b) BuLi, Et_2O , -78 °C (100%); (COCl)₂, DMSO, Et_3N , CH_2Cl_2 , -78 °C to 0 °C (85%); c) PdCl₂, CuCl, O_2 , H_2O , DMF, r.t. (62%); d) LHMDS, THF, -78 °C then MeI, HMPA (91%)

Scheme 3.

Table 1.

Entry	Base	Solvent	Temp	Yield of 2/%	Yield of 14/%
1	NaOMe	MeOH	r.t.	21	-
2	KO ^t Bu	^t BuOH	30 °C	5	-
3	LDA	THF	-78 °C to 0 °C	16	44
4	LDA	THF	0 °C to r.t.	31	44
5	LHMDS	THF	-23 °C to r.t.	29	57
6	LTMP	THF	-23 °C to r.t.	35	59
7	LDA-CeCl ₃	THF	-23 °C to r.t.	64	15

Next, reaction conditions of cyclization of the precursor 3 were screened as shown in Table 1. It revealed that lithium amides promoted this cyclization smoothly to produce desired bicyclic compound 2⁸ in moderate yields (Entry 3 - 6); however, undesired aldol 14 was also produced under these reaction conditions. On the other hand, elimination of t-butyldimethyl-silanol proceeded rapidly and the enone 2 was obtained in low yield when NaOMe or KOtBu was used as a base in a protic solvent. Finally, the best yield was obtained under mild reaction

condition when LDA-CeCl₃ system was employed (Entry 7).⁹

Thus, convenient synthesis of the enone 2, the AB ring model system of taxol, was achieved successfully via allylation of 8-membered ring ketone 6b and intramolecular aldol condensation. A construction of fully functionalized AB ring system of taxol from 8-membered ring enone 1 by the use of above synthetic route II is now in progress.

References and Notes

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- 2; ¹H NMR (CDCl₃) δ = 0.10 (3H, s), 0.12 (3H, s), 0.88 (9H, s), 0.95 1.11 (1H, m), 1.12 (3H, s), 1.29 (3H, s), 1.42 2.01 (7H, m), 1.83 (3H, s), 2.33 2.71 (2H, m), 2.52 (1H, d, J = 18.8 Hz), 2.93 (1H, d, J = 18.8 Hz); ¹³C NMR (CDCl₃) δ = -2.06 (CH₃, TBS), -1.83 (CH₃, TBS), 12.72 (CH₃), 18.35 (C, ^tBu), 22.07 (CH₃), 25.77 (CH₃*3, TBS), 26.11 (CH₂), 28.20 (CH₂), 28.75 (CH₂), 28.99 (CH₂), 30.95 (CH₃), 44.42 (C), 45.16 (CH₂), 50.30 (CH₂), 79.86 (C), 131.81 (C, enone), 167.31 (C, enone), 198.96 (C, enone); IR (neat) 1660 cm⁻¹; MS (EI) 279 (M⁺-^tBu).
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